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Smelting of Low-Silicon Pig Iron in Blast Furnace

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Synopsis :

On the basis of findings of Si content changes in pig iron produced in recent years at blast furnace of Chiba and Mizushima Works, the methods of smelting a low Si pig iron in blast furnace are discussed from the viewpoints of chemical reaction kinetics and liquids flow in the packed bed. Relations of blast furnace operational variable and characteristics to Si contents in pig iron are analyzed using actual data. With a possible target of Si content in pig iron in the near future proposed to be around 0.1%, the paper also discusses some problems which must be solved in attaining the target with a stable furnace operation.

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**The body can be viewed from the next page.**

## Smelting of Low-Silicon Pig Iron in Blast Furnace\*

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*On the basis of findings of Si content changes in pig iron produced in recent years at blast furnace of Chiba and Mizushima Works, the methods of smelting a low Si pig iron*

*blast furnace are discussed from the point of view of the following items:*

*Residue Rem in the packed bed. Deletion of H<sub>2</sub>O from the gas phase.*

[Redacted]

reduction in heat loss from the furnace wall, and  
improvement in efficiency of furnace for CO gas

[Redacted]

$P_{\text{SiO}}$  and  $P_{\text{CO}}$ : Partial pressure of SiO and CO, respectively

$a_{\text{Si}}$ : Activity of Si in pig iron

Eq. (2) holds for a unit volume within the furnace. On the other hand, SiO in coke ash is supposed to

the slag-metal reaction in the dropping zone. At present, the reduction of  $\text{SiO}_2$  contained in coke ash to Si in pig iron via SiO is regarded as a greater factor for the determination of Si content in pig iron than the reduction of  $\text{SiO}_2$  contained in the ore as gangue

depends not only upon the height of dropping zone, but also upon the actual velocity of liquid descent.

cases, its contribution to  $H_s$  should be taken into consideration.

volumetric velocity,  $U_b$ , and total hold-up,  $H_b$ , as in eq. (6).

affected by  $U_M$  and  $U_S$ , respectively, but retention time is largely affected by  $U_M$  and  $U_S$  respectively

$$V_t = \frac{U_t}{U} \dots \dots \dots (6)$$

as it turns short inversely proportional to the increase in  $U_M$  and  $U_S$  because of the correlation indicated in

Since  $U_t$  is the volume of liquid passing through

If the particle size of coke in the dropping zone is

$\Delta s$  for the reaction of SiO generation through the dropping zone are therefore estimated to be in order

contact of SiO<sub>2</sub> in slag with coke,  $A_{sc}$  represents the of  $D_p^{-0.7}$  and  $U_s^{0.4}$ , respectively. The correlation of

#### 4 Features of Blast Furnace Operation vs. Si Content in Pig Iron

##### 4.1 Short-time Relationship

moment to moment. Fig. 5 shows changes in time of  $H_0$  and Si content, and Fig. 6 changes in the correlation coefficient with respect to delayed time between  $H_0$  and Si. Figs. 5 and 6 show that the Si content at

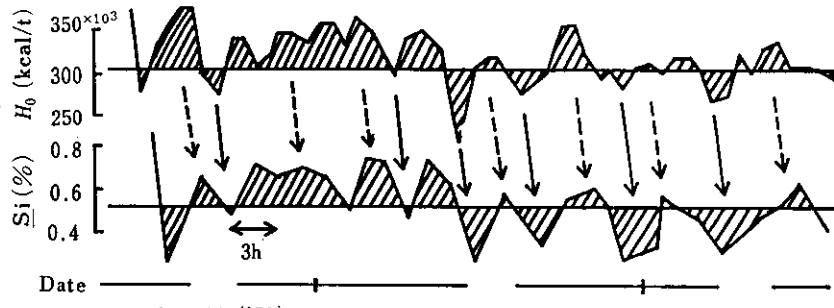


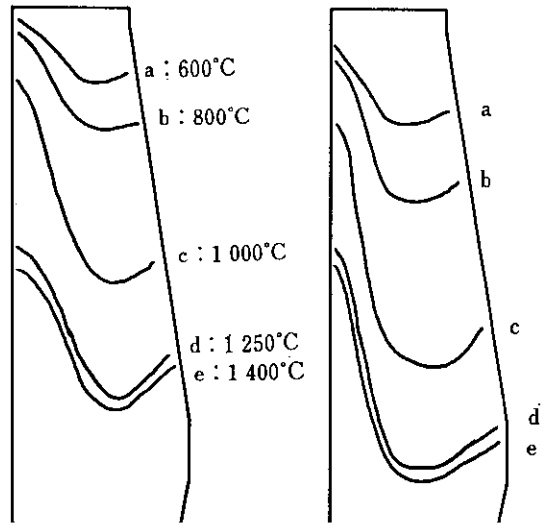
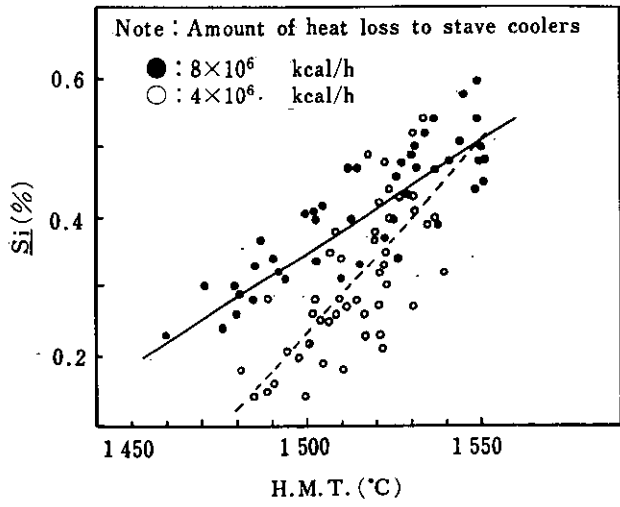
Fig. 5 An example of variation of  $H_0$  and Si content with time (Chiba No. 6 BF)

Inter-correlation coefficient	0.5	Max : 0.47**
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#### 4.2 Long-time Relationship

Fig. 2. 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 12. 13. 14. 15. 16. 17. 18. 19. 20. 21. 22. 23. 24. 25. 26. 27. 28. 29. 30. 31. 32. 33. 34. 35. 36. 37. 38. 39. 40. 41. 42. 43. 44. 45. 46. 47. 48. 49. 50. 51. 52. 53. 54. 55. 56. 57. 58. 59. 60. 61. 62. 63. 64. 65. 66. 67. 68. 69. 70. 71. 72. 73. 74. 75. 76. 77. 78. 79. 80. 81. 82. 83. 84. 85. 86. 87. 88. 89. 90. 91. 92. 93. 94. 95. 96. 97. 98. 99. 100.





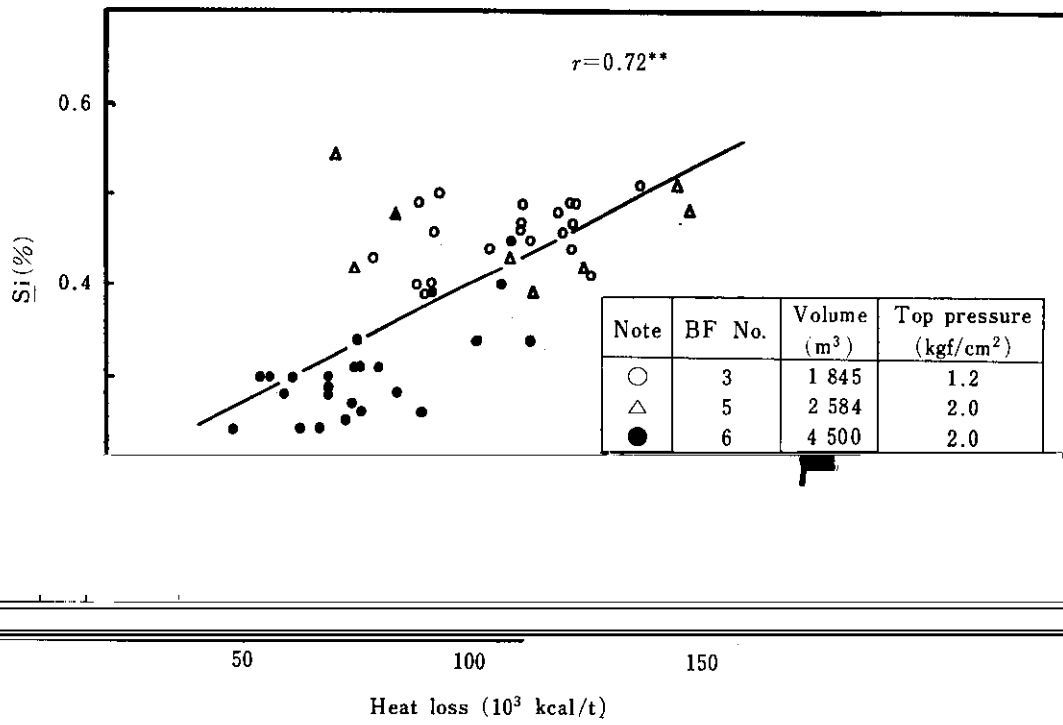


Fig. 17 Influence of heat loss from high temperature region above 950°C on Si content in pig iron (Chiba Works)

Interpolated on the basis of heat balance in the 10 day averages

$$k_f = 4.77 \times 10^8 \exp\left(\frac{-66,500}{RT}\right) \dots\dots(16)$$

and controlling the temperature of coke in the dropping zone while maintaining the currently available

operational techniques of reducing the coke temperature

challenged, aiming at a hot metal temperature of

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